SUPPORTING INFORMATION

An antibiotic efficacy comparative study between market-prescribed and liposomal-based fluoroquinolone ophthalmic nanoformulation against conjunctivitis

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Methodology

Release Kinetics:

The release kinetics of BSF DS from optimized BSF@SLNPs ophthalmic DP was identified by applying various release kinetics equations such as zero-order release kinetics, first-order release kinetics, and the Higuchi model on BSF drug release profile. The released obtained data were calculated by using various parameters. The parameters "n" and time component "k" the release rate constant, and "r" the regression coefficient was determined by the Korsmeyer-Peppas equation to understand the release mechanism.

Result:

The release kinetic results of optimized BSF@SLNPsA9 ophthalmic DP in different equations exhibited different mechanisms. The release constants, the correlation coefficients (r^2) and exponents (n) values of different formulations are shown in **Table S7**. The release model of BSF DS from BSF@SLNPsA9 ophthalmic DP is well fitted with a zero-order model $(r^2: 0.992)$, first-order model $(r^2: 0.975)$, Hixon-Crowell $(r^2: 0.967)$, Higuchi's model $(r^2: 0.987)$ and Korsmeyer-Peppas model $(r^2: 0.991)$ and (n: 1.163). This revealed that the release of the drug was dependent on the encapsulated drug concentration.

Discussion:

The quantitative results obtained *in vitro* release assay are converted by using the following mathematical models [1]. The released data were plotted according to the following equations [2,3].

i. Zero-order model

The equation explains the collective amount of drug release is directly proportional to time. The equation for the zero-order model is mentioned below:

$$C_t = C_0 + K_0 t$$

In the above Equation, where C_t represents the amount of active agent released during the time *t*; C_0 is the initial concentration of active released (generally, $C_0 = 0$); and K_0 is the zero-order constant.

ii. First-order model: The drug release behaviour of the first-order equation presented by the remaining drug (as log cumulative percentage) versus time. The equation for the first-order model is mentioned below [4].

$$\log Q1 = \log Q0 + \frac{k1t}{2.303}$$

In the above equation, where Q_1 is the amount of active agent released on time *t*; Q_0 is the initial amount of drug dissolved; and K_1 is the first-order rate constant.

iii. Hixson-Crowell model: This model depicts the drug release from formulations for the changing diameter of particles and surface area. The equation of the Hixson-Crowell model is mentioned below [5].

$$Q_0^{1/3} - Q_t^{1/3} = K_{HC}$$

In the above equation, Q_t is the drug released amount in t time, Q_0 is the drug amount at an initial time and K_{HC} is the rate constant.

iv. Higuchi model: The Higuchi model is explained as drug released in cumulative percentage versus square root of time. The equation of the Higuchi model is mentioned below [6].

In the above equation, Q is the drug released amount in t time and "K \sqrt{t} " is the rate constant.

v. Korsmeyer-Peppas model: The dissolution results can be fitted to the Korsmeyer-Peppas equation which is used to explain the drug release pattern from the polymer-contained system. The equation of the Korsmeyer-Peppas model is mentioned below:

$$\log (Mt/Mf) = \log k + n \log t$$

In the above equation, M_t is the drug released amount in t time; M_f is the drug release amount at an infinite time; k is the rate constant; n is the diffusion exponent indicative of the mechanism of drug release [7].

References:

- [1] Kumar P, Ganure AL, Subudhi BB, Shukla S. Design and comparative evaluation of invitro drug release, pharmacokinetics and gamma scintigraphic analysis of controlled release tablets using novel pH-sensitive starch and modified starch-acrylate graft copolymer matrices. Iran J Pharm Res 2015;14:677–91.
- [2] Hossain MA, Alam S, Paul P. Development and evaluation of sustained release matrix tablets of Indapamide using methocel K15M CR. J Appl Pharm Sci 2013;3:85–90. https://doi.org/10.7324/JAPS.2013.3516.
- Bruschi MLBT-S to M the DR from PS, editor. Mathematical models of drug release.
 Strategy to Modify Drug Release from Pharm. Syst., Woodhead Publishing; 2015, p. 63–86. https://doi.org/10.1016/b978-0-08-100092-2.00005-9.
- [4] Wagner JG. Interpretation of percent dissolved-time plots derived from in vitro testing of conventional tablets and capsules. J Pharm Sci 1969;58:1253–7. https://doi.org/10.1002/jps.2600581021.
- [5] Hixson AW, Crowell JH. Dependence of Reaction Velocity upon Surface and Agitation:
 I—Theoretical Consideration. Ind Eng Chem 1931;23:923–31. https://doi.org/10.1021/ie50260a018.
- [6] Higuchi T. Rate of release of medicaments from ointment bases containing drugs in suspension. J Pharm Sci 1961;50:874–5. https://doi.org/10.1002/jps.2600501018.
- Korsmeyer RW, Gurny R, Doelker E, Buri P, Peppas NA. Mechanisms of solute release from porous hydrophilic polymers. Int J Pharm 1983;15:25–35. https://doi.org/10.1016/0378-5173(83)90064-9.

List of Supporting Information Tables

Drug substance Name	Besifloxacin hydrochloride							
IUPAC name	7-[(3 <i>R</i>)-3-aminoazepan-1-yl]-8-chloro-1-cyclopropyl-6-fluoro-4- oxoquinoline-3-carboxylic acid; hydrochloride							
Pharmacological Classification	Anti-Bacterial Agents							
Molecular Weight	430.30							
Molecular structure								
Molecular Formula	C19H22Cl2FN3O3							

Table S1: Physico-chemical properties of Besifloxacin hydrochloride drug substance.

Table S2: Details of approved drug product of Besivance (Besifloxacin ophthalmic suspension0.6%) available on electronic Orange Book.

Marketed status	Rx.
Active ingredient	Besifloxacin hydrochloride
Proprietary Name	BESIVANCE
Application Number	N022308
Dosage Form	Suspension/Drops
Route of Administration	Ophthalmic
National Drug Code (NDC)	24208-446-05
Strength	EQ 0.6% BASE (6 mg/mL)
Reference Listed Drug	Yes
Reference Standard	Yes
TE Code	None
Product Number	001
Approval Date	May 28, 2009
Applicant Holder Full Name	BAUSCH AND LOMB INC
Marketing Status	Prescription
Storage	Store at 15°C to 25°C (59°F to 77°F). Protect from light.

Batch no.	SLNPs-1	SLNPs-2	SLNPs-3			
Ingredients	Quantity (mg/mL)					
Besifloxacin HCl	6.6	6.6	6.6			
Glyceryl monostearate	3.3	-	-			
Glyceryl palmitostearate	-	3.3	-			
Glyceryl Behenate	-	-	3.3			
Polysorbate 80	0.5	0.5	0.5			
Poloxamer 188	0.5 0.5		0.5			
Purified water	1 1		1			
	Process parame	eters				
HSH	10000 RPM for10 min during the mixing lipid phase					
HSH	10000 RPM for10 min during the mixing lipid phase					
НРН	5 Nos. cycles at 1000 bar pressure					
Centrifugation	10000 RPM for 10 m	iin				

Table S3: Composition details and process parameters for manufacturing and optimization ofbesifloxacin hydrochloride drug substance-loaded solid lipid nanoparticles (BSF@SLNPs).

Table S4: 3² factorial designs of besifloxacin hydrochloride drug substance-loaded solid lipid nanoparticles (BSF@SLNPs).

Factors (independent variables)	Actual Levels used				
Factors (independent variables)	Low (-1)	Medium (0)	High (+1)		
X1: Ratio of the drug: Lipid	1:1	1:0.75	1:0.5		
X ₂ : HPH cycles at 1000 bar pressure	5	20			
Response (dependent variables)		Limits			
Response 1 (Y ₁): M _D (nm)		$200 \leq Y1 \leq 500$			
Response 2 (Y ₂): ZP (mV)	-20 ≤Y2≤ -60				
Response 3 (Y ₃): % EE		40≤Y3≤80			

Table S5: Composition and process optimization of besifloxacin hydrochloride drugsubstance-loaded solid lipid nanoparticles (BSF@SLNPs) drug product.

Batch no.	SLNPs-4	SLNPs-5	SLNPs-6	
DS and lipid ratio	1:1	1:0.75	1:0.5	
Ingredients		Quantity (mg/mL)		
BSF	6.6	6.6	6.6	
GMS	6.6	-	-	

	-			4.95			-			
	-				-			3.3		
Polysorbate 80	0.5			0.5			0.5			
Poloxamer 188		0.5			0.5			0.5		
WFI		1 mL			1 mL			1 mL		
	SLNPs-	SLNPs-	SLNPs-	SLNPs-	SLNPs-	SLNPs-	SLNPs-	SLNPs-	SLNPs-	
Sub-batch no	4 a	4b	4c	5a	5b	5c	6a	6b	6c	
	Process parameters									
HSH	10,000 aqueou	10,000 RPM for 10 mins during the mixing lipid phase and mixing of the aqueous to lipid phase.								
No. of HPH cycles at 1000 bar pressure	5	10	20	5	10	20	5	10	20	
Centrifugation		10000 RPM for 10 mins								

Abbreviation: BSF: Besifloxacin hydrochloride; GMS: Glyceryl monostearate (C₂₁H₄₂O₄); HPH: High-pressure homogenization; HSH: High-speed homogenization; SLNPs: Solid lipid nanoparticles. **Table S6:** Composition and its quantity for preparation of optimized BSF@SLNPs ophthalmic

 drug product.

Sr. No.	Ingredients	Quantity (mg/mL)
1.	Besifloxacin Hydrochloride (BSF)	6.6
2.	Glyceryl monostearate (GMS)	3.0
3.	Gellan Gum (Gelrite)	3.00
4.	Benzalkonium Chloride	0.10
5.	Mannitol	10.00
6.	Poloxamer 188	0.5
7.	Polysorbate 80	0.5
8.	Sodium Chloride (NaCl)	5.00
9.	Di Sodium Edetate	1.00
10	Sodium Hydroxide (NaOH)	Q.s. to adjust pH
11.	Water for injection (WFI)	Q.s. to 1 mL

Table S7: Result of in vitro release kinetics result for model fitting of optimized BSF@SLNPsA9 ophthalmic

 drug product into simulated tear fluid.

	Zero-order		First order H		Hixon-Crowell		Higuchi		Korsmeyer-Peppas		
Batch no.	Ко	2	K 1	2	K _{HC}	2	K _H	2		2	K _{KP}
	(h ⁻¹)	<i>r</i> -	(h ⁻¹)	<i>r</i> ² ((h ^{-1/3})	<i>r</i> -	(h ^{-1/2})	<i>r</i> -	n	<i>r</i> -	(h -n)
BSF@				 .	T 0 40	0 0 -	0.044	-		0.004	0.044
SLNPsA9	0.213	0.992	-35.113	0.975	-5.049	0.967	0.046	0.987	1.163	0.991	0.066

Table S8: Various ingredients, their concentration and pH of the media used for the preparation of agar media for testing of antimicrobial activity against the *S. aureus* bacterium.

Sr. No.	Ingredients	Concentration (g/liter)
1.	Casein enzymic hydrolysate	15.00
2.	Peptic digest of soya bean meal	5.00
3.	Agar	15.00
4.	Sodium chloride	15.00
5.	pH at 25 °C	7.4 ± 0.2